

# Efficient CO<sub>2</sub> and H<sub>2</sub>O Removal with Novel Adsorbents for Life Support Applications on Mars, Phase I

Completed Technology Project (2018 - 2019)



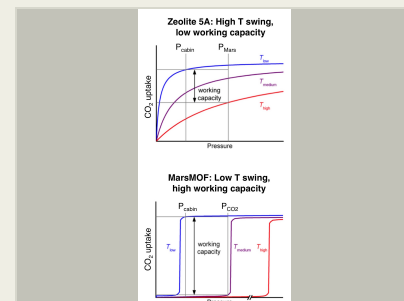
## Project Introduction

Very low CO<sub>2</sub> concentrations that accumulate quickly from human respiration can have dramatic health effects, and thus in NASA's history many technical removal strategies for CO<sub>2</sub> from a confined atmosphere have been suggested and explored. A CO<sub>2</sub> removal system that functions in a is a new area of research, as the primary CO<sub>2</sub> removal component in the state-of-the-art system doesn't have the adsorption performance behavior necessary to function in a Martian atmosphere. We propose to use an alternative adsorbent with unique and highly applicable CO<sub>2</sub> adsorption properties - a diamine-appended metal-organic framework (MOF) - as a drop-in replacement for Zeolite 5A, the CO<sub>2</sub> adsorbent onboard the ISS. Importantly, the mechanism for CO<sub>2</sub> adsorption is disparate from the water adsorption mechanism, allowing the material to be the foundation of newly efficient CO<sub>2</sub> removal processes.

## Anticipated Benefits

Carbon dioxide (CO<sub>2</sub>) removal for breathing life support will always be a necessary component to human NASA missions. This proposal validates a new class of materials with remarkable CO<sub>2</sub> removal properties at the low partial pressures relevant to human toxicity. Additionally, these materials are uniquely suited to perform in a Mars atmosphere. The chemistry of CO<sub>2</sub> removal is more challenging on Mars than in a space vacuum, making the results applicable to any future Mars or non-Mars mission.

Industrial chemistry is a bedrock of modern society. Making chemicals, such as making the ethylene in polyethylene (grocery) bags, has two required components: making the chemical and separating it from any other molecules from the process. While this separation step could seem like an afterthought, in fact it is responsible for at least 10% of global industrial energy use! MarsMOF and the class of materials it belongs to have the potential to improve the cost of the world's largest separations.



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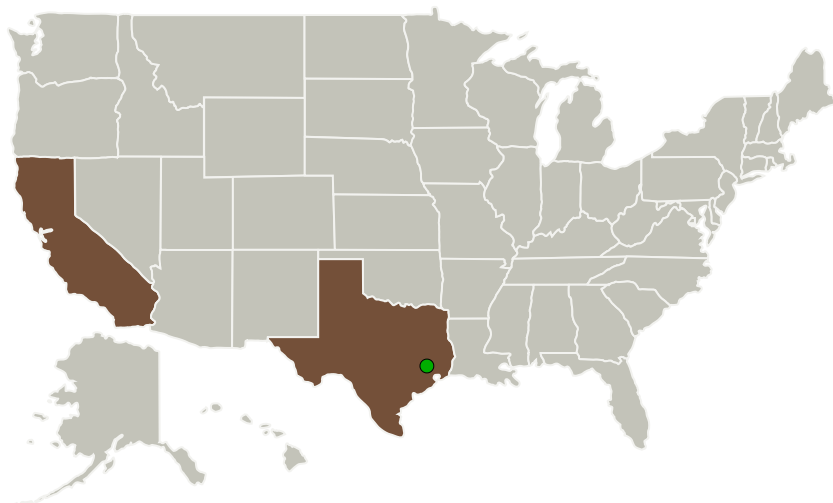
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## Primary U.S. Work Locations and Key Partners



Organizations Performing Work	Role	Type	Location
Mosaic Materials, Inc.	Lead Organization	Industry	Berkeley, California
● Johnson Space Center(JSC)	Supporting Organization	NASA Center	Houston, Texas
University of California-Berkeley(Berkeley)	Supporting Organization	Academia	Berkeley, California

## Primary U.S. Work Locations

California	Texas
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## Project Transitions

**July 2018:** Project Start

**August 2019:** Closed out

## Closeout Documentation:

- Final Summary Chart(<https://techport.nasa.gov/file/141331>)

## Organizational Responsibility

## Responsible Mission Directorate:

Space Technology Mission Directorate (STMD)

## Lead Organization:

Mosaic Materials, Inc.

## Responsible Program:

Small Business Innovation Research/Small Business Tech Transfer

## Project Management

## Program Director:

Jason L Kessler

## Program Manager:

Carlos Torrez

## Principal Investigator:

Thomas M McDonald

## Technology Maturity (TRL)

Start: 3

Current: 4

Estimated End: 4

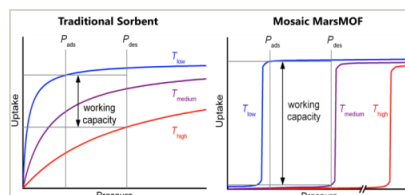
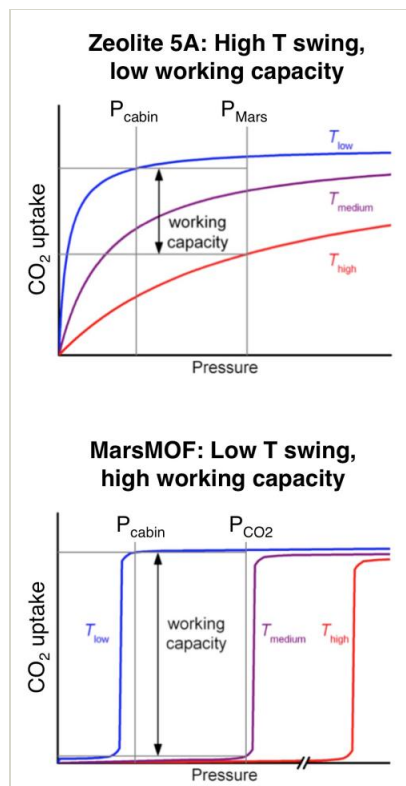


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## Images



### Final Summary Chart Image

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(<https://techport.nasa.gov/image/136309>)

### Briefing Chart Image

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(<https://techport.nasa.gov/image/128580>)

## Technology Areas

### Primary:

- TX06 Human Health, Life Support, and Habitation Systems
  - ↳ TX06.1 Environmental Control & Life Support Systems (ECLSS) and Habitation Systems
    - ↳ TX06.1.1 Atmosphere Revitalization

## Target Destinations

Earth, Mars